

MONKEYPOX DISEASE TREATMENT BY TECOVRIMAT ADSORBED ONTO SINGLE-WALLED CARBON NANOTUBE THROUGH DRUG DELIVERY METHOD

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ABSTRACT

In this work, it has been investigated the Monkeypox disease which occurs in both humans and animals by infection with a double-stranded DNA virus having the symptoms consisting of fever, fatigue, headache, and muscle pains like flu. Tecovirimat drug can be applied in prohibition of monkeypox virus through adsorbing onto surface of single-walled carbon nanotube (SWCNT) as the drug delivery method due to direct electron transfer principle which has been studied by density functional theory (DFT) methods.

Therefore, it has been accomplished the B3LYP/6-311+G (2d,p) level of theory to estimate the susceptibility of SWCNT for adsorbing tecovirimat through nuclear magnetic resonance and thermodynamic parameters. In other words, the data explained that the feasibility of using SWCNT and tecovirimat becomes the norm in drug delivery system which has been achieved by quantum calculations due to physico-chemical properties of NMR and IR methodologies.

Keywords: Monkeypox disease; tecovirimat; (5,5) armchair CNT; drug delivery; DFT; IR; NMR.

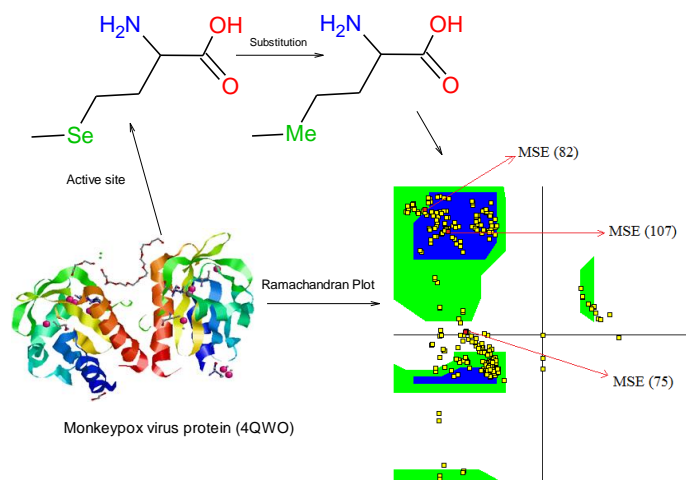
1. INTRODUCTION

One of the recent infectious diseases is monkeypox which might happen in human body and some of the animals with several symptoms such as fever, muscle pain, inflamed lymph nodes, and a pimple that causes smallpox and then crusts disappear [1]. About two to four weeks take time to see the symptoms which might be weak, but sometimes may be severe, particularly in children and pregnant women or people with weak immune systems [1-3].

The disease which can appear like chickenpox is caused by monkeypox viruses (a kind of Orthopoxvirus) and might be scattered from bushmeat, animal scratches or bites, fluids of body, infected compounds or infected people through small droplets and probability the air path [4,5].

Although there is no treatment for monkeypox disease, the smallpox vaccine can be effective in prohibiting infection in close contacts and in decreasing the intensity of the disease. During prevalence, some antiviral drugs, tecovirimat and cidofovir, vaccinia immune globulin and the smallpox vaccine might be applied [6,7].

In 2014, Minasov and co-workers have discovered the crystal structure of A42R Profilin-like protein from Monkeypox Virus using X-ray diffraction [8]. It was seen that amino acid of serine in the target of L-PEPTIDE LINKING with formula C₅H₁₁N O₂Se from the A42R Profilin-like protein is substituted with methionine which has been shown by MSE abbreviation in the Ramachandran plot using VMD (1.9.4) program package (Scheme1) [9].



Scheme1. 3-D structure of Monkeypox virus protein (4QWO) by raswin software as Ribbon view, substitution of serine with methionine in the target and Ramachandran plot using VMD (1.9.3) program package. The blue, green, and white regions represent the favored, allowed, and disallowed or outlier regions with yellow data points of the amino acid sequence.

The Ramachandran plot for monkeypox virus in Scheme1 has depicted the yellow data points from a large set of high-resolution structure and contour for favored region (blue zone), allowed region (green zone) and disallowed or outlier region (white zone). These substituted residues of MSE including MSE (75), MSE (82), and MSE (107) in two chains of A and B with different angles of amino acids (Phi, Psi) causing monkeypox virus have been reported in Table 1 and indicated in Scheme1.

Table1. Residues of Monkeypox virus protein (4QWO) introducing the angles of amino acids (Phi, Psi) in A and B chains at Ramachandran plot using VMD (1.9.4) program package.

Chain	MSE (75)		MSE (82)		MSE (107)	
	Phi	Psi	Phi	Psi	Phi	Psi
A	-93.3565	3.3448	-142.2137	152.9301	-114.6429	125.2847
B	-77.5479	-16.8932	-132.0752	144.5321	-116.2262	128.5598

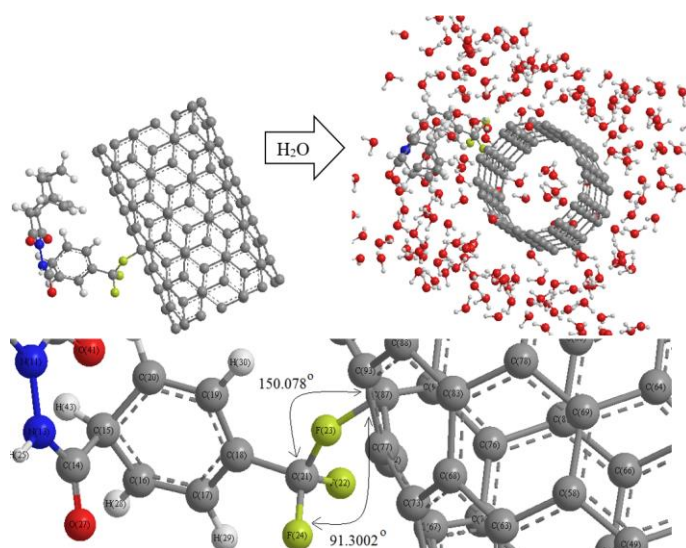
In this work, we have investigated the characteristics of **tecovirimat**, an antiviral drug, which is employed to treat smallpox, cowpox and monkeypox. Tecovirimat, C₁₉H₁₅F₃N₂O₃, is an inhibitor of the orthopoxvirus protein which is exhibited for the cure of human smallpox disease. The impact of tecovirimat might decrease in patients having an impaired immune system [10].

There is an attention to enhancing the bioavailability and duration of action for a drug to modify therapeutic consequences. Drug delivery technique is able to change a drug's pharmacokinetics and specificity by formulating it with various ingredients, drug carriers, and the medical equipment [11-18].

Nanomedicine in drug delivery is for achieving the improved delivery of water insoluble drugs, delivery of large macromolecule drugs to intracellular sites of action, and codelivery of two or more drugs or therapeutic agents for combination remedy [19-23].

Nanotubes with their intrinsic properties have been considered potential candidates for drug delivery carriers. The capped ends of nanotubes may be opened up by oxidation, allowing for the insertion of molecules of interest inside the nanotube. Carbon nanotubes (CNTs) can easily penetrate cells, delivering drugs directly to the cytoplasm or nucleus. Nanotubes conform to a perpendicular position with the cell membrane during uptake, perforating and diffusing through the lipid bilayer to enter the cytoplasm. Functionalized CNTs are easily internalized by cells through passive and endocytosis-independent mechanisms [24-32].

In this research, we have focused on tecovirimat drug adsorbed onto the surface of (5,5) armchair SWCNT in water medium for preventing the activity of monkeypox virus (Scheme2).



Scheme2. Adsorption of tecovirimat onto the surface of (5,5) armchair SWCNT in water medium.

The tecovirimat structure has been investigated in this study as a relatively stable drug for adsorption onto the surface of (5,5) armchair SWCNT through drug delivery method (Scheme2) [33,34].

Thus, a series of quantum theoretical approaches including DFT methods has been accomplished for finding the optimized coordination of [tecovirimat-(5,5) armchair SWCNT] complex using Gaussian 16 revision C.01 program (Scheme2) [35].

2. THEORETICAL BACKGROUND, MATERIAL AND METHODS

In this study, the geometry coordination has been optimized at the framework of DFT using the three-parameter Becke's exchange [36] and Lee-Yang-Parr's correlation non-local functional [37], usually known as B3LYP method and basis set of 6-311+G(2d,p). The density functional theory (DFT) is one of the most employed approximations of Hohenberg, Kohn and Sham which allows the theoretical study of material properties [38]. DFT theory proves an advantageous method for predicting chemical systems, and in order to understand its similarities and differences to other computational methods employed [39-41].

Then, it has been described the electronic structure of adsorbed (5,5) armchair SWCNT by tecovirimat for measuring physico-chemical properties (Scheme2).

In this investigation, the Onsager model has been accomplished that was developed by Frisch, Wong and Wiberg utilizes spherical cavities. Even though this implies a less accurate description of the solute-solvent interface, this approximation simplifies the evaluation of energy formatives in geometry optimizations, and frequency analysis. Moreover, Cramer and Truhlar improved this model at the dipole level [42-46]. In fact, a cavity must have a physical sense such as Onsager model, and has a mathematical ability as often happened in other descriptions of solvent impacts [47]. On the other hand, the cavity has to keep out the solvent and including its frontiers as the biggest probability part of the solute charge distribution [47-50].

Basically, a group of quantum theoretical methods has been run for exploring some physical and chemical properties from optimized structure of tecovirimat adsorbed onto the surface of (5,5) armchair SWCNT including charge distribution, thermodynamic calculations and nuclear magnetic resonance analysis due to designing a drug delivery model using Gaussian 16 revision C.01 program [51]. Moreover, the gauge including atomic orbitals (GIAO) has been adopted to solve the gauge problem in the calculation of nuclear magnetic shielding for [tecovirimat-(5,5) armchair SWCNT] complex using density functional theory (DFT) calculation.

3. RESULTS AND DISCUSSION

Carbon nanotubes are remarking drug delivery platforms that can be functionalized with various biomolecules including antibodies, proteins, and DNA. This permits the particular target for transferring the special tissues,

organs, or cells. These compounds can easily penetrate cells, delivering drugs directly to the cytoplasm or nucleus. Drug delivery systems improve the pharmacological and therapeutic profile and efficacy of the drug and lower the occurrence of off-targets [52].

3.1. Charge transfer

In table2, the atomic charge, spin density and Isotropic Fermi Contact Couplings (10^{-4} cm^{-1}) of labeled carbon, nitrogen, oxygen, and fluorine have been calculated through optimized [tecovirimat-(5,5) armchair SWCNT] complex in water medium. It has been exhibited that certain atoms in active sites of tecovirimat adsorbed onto the surface of (5,5) armchair SWCNT play an important role for the electron charge transfer, spin density, and Isotropic Fermi Contact Couplings in this structure.

Table2. The atomic charge, spin density, and Isotropic Fermi Contact Couplings (10^{-4} cm^{-1}) of carbon, nitrogen, oxygen, and fluorine for [tecovirimat-(5,5) armchair SWCNT] complex.

Atom	Atomic charge	Spin density	Isotropic Fermi Contact Couplings
N11	-0.2242	0.0004	-0.0198
C12	0.2986	-0.0009	-0.1921
N13	-0.0506	-0.0091	0.16163
C14	0.3222	-0.0112	16.5137
C15	-0.0078	-0.1943	-25.7744
C16	0.0342	1.1223	150.5479
C17	0.0209	-1.0848	-151.8102
C18	-0.0160	1.0653	141.8360
C19	0.0109	-1.0853	-151.4566
C20	0.0240	1.1124	148.6568
C21	0.4819	-0.0664	-11.9191
F22	-0.1778	0.0029	-3.5263
F23	-0.1796	0.0031	-3.5375
F24	-0.1687	0.0343	0.0803
O25	-0.2566	0.0003	-0.0051
O26	-0.2658	0.0877	-2.1655
O27	-0.2564	0.0001	-0.0294

Then, in Figure1a, it has been also plotted and indicated that the most negative atomic charges belong to N11, N13, F23, O25 and O27, while C15, C16, C17, C18, C19, C20 and C21 have shown the most positive and negative fluctuations in spin density and Isotropic Fermi Contact Couplings plots through optimized [tecovirimat-(5,5) armchair SWCNT] complex in water medium (Figure1b,c). Besides, it has been observed the weak changes in the element of Isotropic Fermi Contact Couplings around H37, H38, H39, H40, H41, H42, and H43 in this compound (Figure1c).

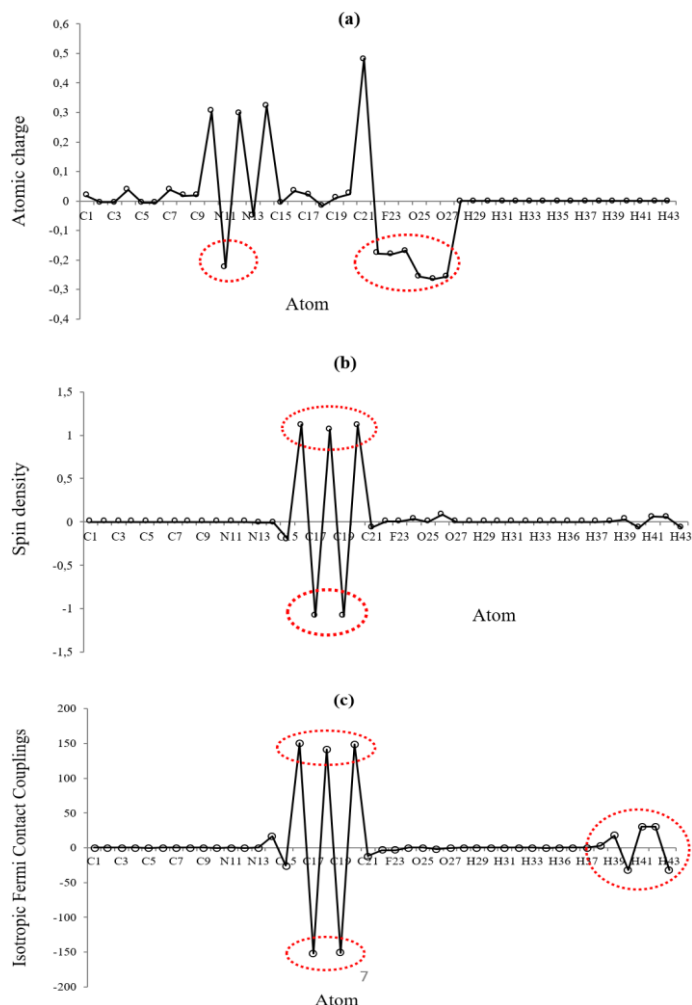


Figure 1. Changes of **a)** atomic charge, **b)** spin density, **c)** Isotropic Fermi Contact Couplings (10^{-4} cm^{-1}) versus labeled hydrogen, carbon, oxygen, nitrogen, and fluorine through optimized [tecovirimat-(5,5) armchair SWCNT] complex in water medium.

So, the results of Table 2 and Figure 1 in a polar medium of water solution indicate the stability of tecovirimat drug joint to the surface of (5,5) armchair SWCNT as a drug delivery technic for treatment the monkeypox disease.

The outlook of Figure 1 recommends the reason for existing observed various results of [tecovirimat-(5,5) armchair SWCNT] complex which are principally bound to the position of active sites of labeled hydrogen, carbon, oxygen, nitrogen, and fluorine which move the charge of electrons in tecovirimat in polar water molecules.

In fact, the partial charges have been obtained by fitting the electrostatic potential to fixed charge of fluorine atoms of tecovirimat adsorbed to the surface of (5,5) armchair SWCNT. Therefore, the electrophilic side chains of tecovirimat conduct us to find the reason for the activity and the stability of this structure which prevent the monkeypox virus activity.

3.2. NMR analysis

The NMR data of isotropic (σ_{iso}), anisotropic shielding tensor (σ_{aniso}), and eigenvalues of chemical shielding including σ_{11} , σ_{22} , σ_{33} (ppm) for tecovirimat adsorbed onto the surface of (5,5) armchair SWCNT, respectively, have been estimated (Table 3).

The computed results have indicated the SCF GIAO Magnetic shielding tensor in ppm for hydrogen, carbon, nitrogen, oxygen and fluorine exploring the active site of tecovirimat compound as the drug for treatment the monkeypox disease.

The calculations have been done based on B3LYP/6-311+G(2d,p) level of theory using Gaussian 16 revision C.01 program [35] and reported in Table 3.

Table 3. SCF GIAO Magnetic shielding tensor (ppm) for tecovirimat adsorbed onto the surface of (5,5) armchair SWCNT in ppm using B3LYP/6-311+G(2d,p) method.

Atom	σ_{11}	σ_{22}	σ_{33}	σ_{iso}	σ_{aniso}
N1	179.0301	196.4202	241.9750	205.8084	54.2498
N13	195.4342	239.3027	296.1563	243.6311	78.7878
C17	58.8329	118.1352	211.7586	129.5755	123.2745
C18	64.4449	113.5200	212.2859	130.0836	123.3035
C19	58.0918	121.6569	211.6560	130.4682	121.7817
F22	423.2038	485.7044	518.1468	475.6850	63.6927
F23	422.8749	486.4392	518.5336	475.9493	63.8765
F24	413.4528	484.3823	488.5692	462.1348	39.6516
O25	-662.4813	-243.3194	485.2842	-140.1722	938.1845
O26	-597.5044	-224.3702	464.0855	-119.2630	875.0228
O27	-655.5859	-241.8773	485.0881	-137.4584	933.8197
H29	25.0471	27.2044	44.8719	32.3745	18.7462
H30	24.0675	28.5044	43.8365	32.1361	17.5506

Optimized energy: $-842.6840 \times 10^{-3} \text{ kcal/mol}$

Dipole moment: 5.0851(Debye)

The [tecovirimat-(5,5) armchair SWCNT] complex has shown the chemical shielding including σ_{11} , σ_{22} , σ_{33} and σ_{iso} , σ_{aniso} (ppm) for various atoms of hydrogen, carbon, nitrogen, oxygen and fluorine in the active sites of the molecule through the NMR graph (Figure 2 a). The most fluctuations have been observed in O25, O26, O27 and F22, F23, F24, respectively (Figure 2 a).

Moreover, the ^{13}C -NMR measurements on tecovirimat have demonstrated the active sites of this compound due to exploring the most electronegative atoms for adsorbed onto the surface of (5,5) armchair SWCNT which represent the maximal shift in TMS B3LYP/6-311+G(2d,p) (Figure 2 b).

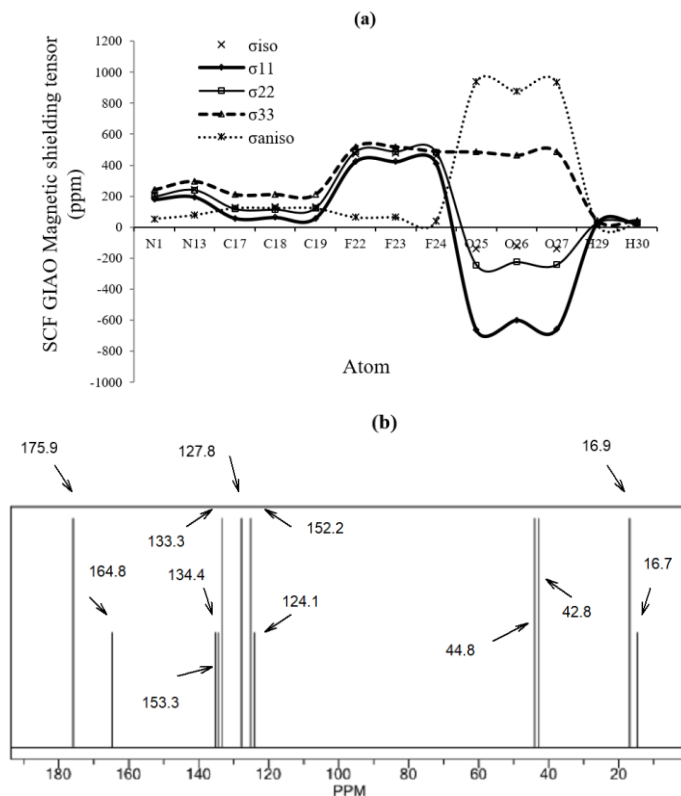


Figure 2. **a)** The NMR plot of isotropic (σ_{iso}), anisotropic (σ_{aniso}) and eigenvalue shielding tensors (σ_{11} , σ_{22} , σ_{33}) calculated by level of theory B3LYP/6-311+G(2d,p) for [tecovirimat-(5,5) armchair SWCNT] complex; **b)** ^{13}C -NMR data on tecovirimat adsorbed onto the surface of (5,5) armchair SWCNT using TMS B3LYP/6-311+G(2d,p) level of theory.

The CS tensors are yielded by the quantum chemical calculations in principal axes system to estimate the isotropic chemical-shielding (CSI) and anisotropic chemical-shielding (CSA) based on the equations 1 & 2 [53-55]:

$$\text{CSI (ppm)} = (\sigma_{33} + \sigma_{22} + \sigma_{11})/3 \quad (1)$$

$$\text{CSA (ppm)} = \sigma_{33} - (\sigma_{22} + \sigma_{11})/2 \quad (2)$$

Besides, the Onsager model has influenced on the nuclear magnetic resonance data and chemical shielding of hydrogen, carbon, nitrogen, oxygen and fluorine atoms in [tecovirimat-(5,5) armchair SWCNT] complex (Figure 2 a,b).

3.2. IR method

The calculations of the infrared (IR) spectrum have been accomplished for tecovirimat adsorbed onto the surface of (5,5) armchair SWCNT using B3LYP method and 6-311+G(2d,p) basis set for atoms including hydrogen, carbon, oxygen, nitrogen, and fluorine to obtain the more accurate equilibrium geometrical parameters, thermodynamic properties and data for each of the determined structure. The IR spectrum for tecovirimat drug has been seen in the frequency range about 300 cm^{-1} - 3650 cm^{-1} (Figure 3). It can be seen that the strongest allowed peaks with highest frequency occur about 1350 cm^{-1} , 1875 cm^{-1} , 1900 cm^{-1} , respectively (Figure 3).

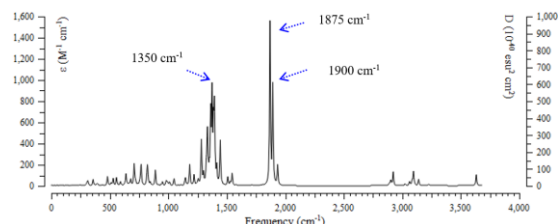


Figure 3. Calculated IR intensity (km/mol) versus frequency (cm^{-1}) through the IR spectrum for tecovirimat drug adsorbed onto the surface of (5,5) armchair SWCNT using 6-311+G(2d,p) calculations.

The perspective of Figure 3 recommends the reason for existing observed various results frequencies of [tecovirimat-(5,5) armchair CNT] complex which present the position of active sites of labeled hydrogen, carbon, nitrogen, oxygen and fluorine in this drug structure that transfer the charge of electrons in polar tecovirimat in water toward the surface of (5,5) armchair carbon nanotube.

The calculations of the relative harmonic frequencies, IR intensities in various normal modes and thermodynamic properties of ΔH , ΔG , ΔS for [tecovirimat-(5,5) armchair CNT] complex using B3LYP/6-311+G(2d,p) method have been reported in Table 4.

Table 4. Calculated functions of harmonic frequencies (cm^{-1}), IR intensities (km/mol) in different normal modes; thermodynamic properties of ΔG , ΔH in kcal/mol and ΔS in cal/mol.K^{-1} at using B3LYP/6-311+G(2d,p) method at 300K.

Compounds	Normal mode	IR Intensity (km/mol)	Frequency (cm^{-1})
[Tecovirimat- onto (5,5) armchair SWCNT]	19	13.5481	312.0300
	20	17.8254	357.0202
	26	25.8903	479.6284
	28	20.3705	528.7465
	30	18.6756	556.6860
	35	30.8494	638.1736
	40	59.7066	707.6418
	45	43.3174	767.8376
	47	46.9313	821.4854
	52	43.2173	888.9225
	72	55.5913	1182.4557
	75	28.3502	1219.9341
	78	123.1229	1282.6031
	82	47.0726	1331.1888
	83	117.1697	1334.7375
	86	182.9788	1362.3063
	88	228.8568	1373.5365
	90	144.8028	1385.3614
	91	154.9904	1393.5278
95	121.3159	1443.8165	
105	444.4583	1868.7818	
106	270.9917	1892.7379	
119	36.1119	3095.3813	
123	30.9159	3632.4809	

$\Delta G \times 10^{-3} = -858.126 \text{ kcal/mol}$, $\Delta H \times 10^{-3} = -858.086 \text{ kcal/mol}$, $\Delta S = 132.879 \text{ cal/K.mol}$, Dipole moment = 7.6616 Debye

Figure 4 demonstrates the changes of frequency versus various harmonic normal modes with sharp peaks for tecovirimat adsorbed onto the surface of (5,5) armchair SWCNT at B3LYP/6-311+G(2d,p) quantum method.

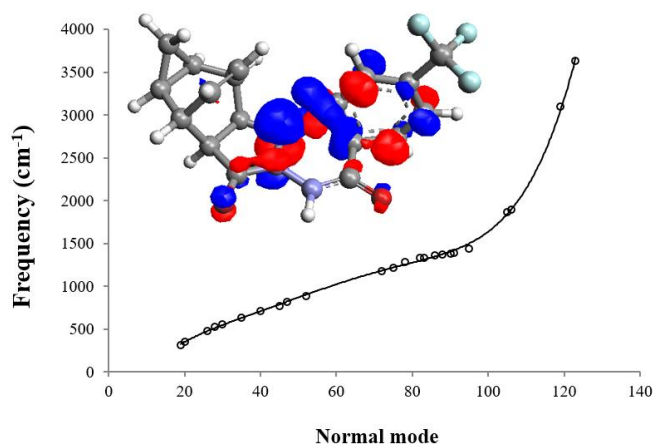


Figure 4. Harmonic frequency (cm^{-1}) versus normal mode for tecovirimat adsorbed onto the surface of (5,5) armchair SWCNT using B3LYP/6-311+G(2d,p) quantum calculation.

It has been notable that polarization functions into the applied basis set in the computations always demonstrate a significant achievement on the simulation and modeling methods of theoretical levels. The normal modes of IR spectrum have been exploring of harmonic potential wells by analytic methods which keep the movement of all atoms at the same time in the vibration time scale leading to a natural definition of molecular vibrations (Figure 4).

The results of the above observations strongly suggest that tecovirimat adsorbed onto the surface of (5,5) armchair SWCNT at B3LYP/6-311+G(2d,p) method in water solvent is predominantly due to basis set functions which are induced by a change in polarity of the environment. It is clear that an increase in the dielectric constant increases the stability and efficiency of this drug for treating monkeypox disease.

CONCLUSION

In this work, monkeypox has studied as a zoonotic poxvirus disease which could happen in human and other animals due to substitution amino acid of serine with methionine. Then, we have investigated the tecovirimat drug for treating monkeypox disease through adsorbing onto the surface of (5,5) armchair SWCNT at B3LYP/6-311+G(2d,p) level of theory in water medium as the drug delivery method.

Tecovirimat has attracted much attention for the clinical treatment of monkeypox disease through adsorption onto the surface of (5,5) armchair SWCNT which introduces an efficient drug delivery system though NMR and IR data analysis on the optimized structure.

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